

Name _____

Drawing Covalent Compounds

<p>1. H_2O</p> <p>a) Total # of outer electrons:</p> <p>b) Compound Name:</p>	<p>c) Drawing</p>
<p>2. NH_3</p> <p>a) Total # of outer electrons:</p> <p>b) Compound Name:</p>	<p>c) Drawing</p>
<p>3. CH_4</p> <p>a) Total # of outer electrons:</p> <p>b) Compound Name:</p>	<p>c) Drawing</p>
<p>4. HCl</p> <p>a) Total # of outer electrons:</p> <p>b) Compound Name:</p>	<p>c) Drawing</p>
<p>5. PCl_3</p> <p>a) Total # of outer electrons:</p> <p>b) Compound Name:</p>	<p>c) Drawing</p>

<p>6. PCl_5</p> <p>a) Total # of outer electrons:</p> <p>b) Compound Name:</p>	<p>c) Drawing</p>
<p>7. NF_3</p> <p>a) Total # of outer electrons:</p> <p>b) Compound Name:</p>	<p>c) Drawing</p>
<p>8. O_2</p> <p>a) total # of outer electrons:</p> <p>b) Compound Name: Diatomic Oxygen</p>	<p>c) Drawing</p>
<p>9. CO_2</p> <p>a) total # of outer electrons:</p> <p>b) Compound Name:</p>	<p>c) Drawing</p>