

NAMING COMPOUNDS

DIRECTONS: Write the names for the following compounds in the space provided. Use Roman numerals where appropriate. Use covalent prefixes where appropriate. Circle the number of the covalent compounds (two nonmetals).

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| 1. KOH _____ | 17. HgO _____ |
| 2. NH ₄ NO ₃ _____ | 18. SO ₃ _____ |
| 3. LiCl _____ | 19. SO ₂ _____ |
| 4. NaC ₂ H ₃ O ₂ _____ | 20. P ₄ O ₁₀ _____ |
| 5. PCl ₃ _____ | 21. Na ₂ CO ₃ _____ |
| 6. N ₂ O ₅ _____ | 22. (NH ₄) ₃ PO ₄ _____ |
| 7. Pb(OH) ₂ _____ | 23. N ₂ O _____ |
| 8. FeCl ₃ _____ | 24. N ₂ O ₃ _____ |
| 9. K ₂ SO ₄ _____ | 25. SiCl ₄ _____ |
| 10. NaClO ₃ _____ | 26. SnI ₄ _____ |
| 11. NH ₄ Cl _____ | 27. Co(NO ₂) ₂ _____ |
| 12. Ba(C ₂ H ₃ O ₂) ₂ _____ | 28. AsCl ₄ _____ |
| 13. (NH ₄) ₂ S _____ | 29. AsCl ₅ _____ |
| 14. CuCl ₂ _____ | 30. LiBr _____ |
| 15. Zn ₃ (PO ₄) ₂ _____ | 31. MgCl ₂ _____ |
| 16. CO ₂ _____ | 32. H ₂ SO ₄ _____ |