## Welcome to AP Physics 1 Mr. Allan

## What's AP Physics all about?

- 2 Semester, algebra-based physics course, equivalent to a 1 st semester college physics class
- At least $25 \%$ of time spent in lab - need to keep a "lab notebook"
- Ruler/Straight Edge
- Calculator
- Course description and details found on AP Classroom (Code: XEX7Q7)
- Mrallansciencegfc.com
- Focus will be on group work, inquiry, and developing understanding together


## AP Physics 1 Test

## Section 1 - MCQs:

- 50 multiple choice questions
- 1 hour 30 minutes
- $50 \%$ of test score

Questions are either discrete questions or question sets, in which students are provided with a stimulus or a set of data and a series of related questions.

Multi-select questions: 5 of the 50 questions have two correct answers - must choose both to get it right

## Section 2 - FRQs:

- 5 free response questions
- 1 hour 30 minutes
- $50 \%$ of test score


## Question Types:

- Experimental Design (1)
- Qualitative/Quantitative Translation (1)
- Short Answer: Paragraph Argument (1)
- Short Answer (2)

Suggested timing will be given

## Course Outline

Unit 1 - Kinematics in 1 and 2 Dimensions

- Measurements; Velocity; Acceleration; Vectors
- 16-19 days
- 12-18\% AP Exam Weighing

Unit 2 - Dynamics

- Forces; Newton's Laws; Friction and Dynamics
- 21-24 days
- 16-20\% AP Exam Weighing


## Unit 3 - Circular Motion and Gravitation

- Uniform circular motion/dynamics; Gravity
- 8-10 days
- 6-8\% AP Exam Weighing


## Course Outline

Unit 4 - Energy and Conservation of Energy

- Work; Power; Kinetic Energy; Potential Energy
- 22-25 days
- 20-28\% AP Exam Weighing


## Unit 5 - Momentum

- Impulse; Momentum; Collisions, Conservation of Momentum
- 14-17 days
- 12 - 18\% AP Exam Weighing


## Course Outline

## Unit 6 - Simple Harmonic Motion

- SHM; Graphing of SHM; Simple Pendulum; Springs
- 4-7 Days
- 4-6\% AP Exam Weighing

Unit 7 - Rotational Motion, Torque, and Angular Momentum

- Torque; Rotational Quantities
- 14-19 Days
- 12 - 18\% AP Exam Weighing


## Units

- Numerical answers do not mean anything unless they are labeled in proper units
- All answers must be labeled in proper units
- We will use different units for 3 primary different types of measurements
- Length
- Time
- mass


## Units

SI Base Units--- the standard unit a quantity is measured in

| Quantity | Base Unit | Symbol |
| :--- | :--- | :--- |
| Length | Meter | m |
| Time | Second | s |
| Mass | Kilograms | $\mathrm{Kg}^{* *}$ |

Metric Prefixes- smaller or bigger divisions of base units

| Name | Symbol | How it relates to <br> base unit |
| :--- | :--- | :--- |
| Kilo- | k | x 1000 |
| Base Unit |  | $\times 1$ |
| Centi- | cm | $\times 1 / 100$ |
| Milli- | m | $\mathrm{x} 1 / 1000$ |
| Micro- | m | $\mathrm{X} \mathrm{1/1,000,000}$ |
| Nano- | n | $\mathrm{X} 1 / 1,000,000,000$ |

## SI Units

- Base Units


## Base SI Units

* Le Systéme Internationale (SI) units are standard in science

| Quantity | Base Unit | Symbol |
| :--- | :--- | :---: |
| length | meter | m |
| mass | kilogram | kg |
| time | second | s |
| temperature | kelvin | K |

## - Derived Units

- Built from base SI units


## Derived Units

* built from base SI units

| area | length $\times$ length | $\mathrm{m}^{2}$ |
| :--- | :--- | :--- |
| velocity | length/time | $\mathrm{m} / \mathrm{s}$ |
| density | mass/volume | $\mathrm{kg} / \mathrm{m}^{3}$ |

* Example: Give derived units for force and for energy.


## Dimensional Analysis

## The Dimensional Analysis method was developed to:

- Can change one set of units to another
- Equalities (i.e., conversion factors) are set up in fraction form
- Equalities lined up sequentially and units used on the top and bottom of neighboring fractions are alternated so that units cancel


## Two steps to problems:

Step 1: State the given quantity (number and units) and unknown
Step 2: Start with what you know

Factor Label Example

## $55.0 \mathrm{~km} / \mathrm{hr}=$ ? m/s

$1 \mathrm{~km}=1000 \mathrm{~m}$
$1 \mathrm{hr}=60 \mathrm{~min}$
$1 \mathrm{~min}=60 \mathrm{sec}$

> - Must cancel units
> -Must show units
> -Box answer

$$
\left(\frac{55.0 \mathrm{kAT}}{1 \mathrm{bH}}\right)\left(\frac{1000 \mathrm{Rt}}{1 \mathrm{~km}}\right)\left(\frac{1 \mathrm{hr}}{60 \mathrm{mirt}}\right)\left(\frac{1 \mathrm{miK}}{60 \mathrm{sec}}\right)=15.3 \mathrm{~m} / \mathrm{sec}
$$

## Significant Figures

## Certain

All three numbers are important.

## There are three "Siqnificant Fiqures".



## Significant Figure Rules

Nonzero integers always count as sig figs

- 3456 has 4 sig figs

Leading zeros do not count as sig fig

- 0.0486 has 3 sig figs

Captive zeros always count as sig fig

- 16.07 has 4 sig fig

Trailing zeros are only significant only if the number contains a decimal point

- 9.300 has 4 sig fig
- 9300 has 2 sig fig

| 2.47 kilometers |  | 8.2 millimeters |
| :---: | :---: | :---: |
| 3 sig figs |  | 2 sig figs |
| 2,470 meters |  | 0.0082 meters |
| 3 sig figs |  | 2 sig figs |


| $2,470,000$ millimeters | 0.0000082 kilometers |
| :---: | :---: |
| 3 sig figs | 2 sig figs |

## Zeros are NOT "significant" when they are merely place holders.

## Sig Fig Rules

## Multiplication \& Division

- The value with the fewest sig figs determines the number of sig figs in the answer
- Least amount

$$
6.38 \times 2.0=12.76
$$

$$
\text { = } 13 \text { (2 sig figs) }
$$

## Addition \& Subtraction

- The number of decimal places in the result equals the number of places in the least precise measurement
- Least precise (poorest measurement)
$6.8+11.934=18.734$
$=18.7$

$$
\begin{gathered}
\text { How many } \frac{\boldsymbol{m i}}{\boldsymbol{h r}} \text { is } 42.5 \frac{\boldsymbol{m}}{\boldsymbol{s}} ? \\
m \rightarrow k m \rightarrow m i \\
s \rightarrow \min \rightarrow h r \\
\frac{42.5 \pi}{s} \frac{1 \mathrm{~km}}{1000 \pi} \frac{1 \mathrm{mi}}{1.609 \mathrm{~km}} \frac{6 \infty \mathrm{~s}}{1 \mathrm{~m}} \frac{60 \mathrm{~min}}{1 \mathrm{hr}}
\end{gathered}
$$



$$
\begin{aligned}
& \frac{42.5 \cdot 60 \cdot 60 \mathrm{mi}}{1000 \cdot 1.609 \mathrm{hr}} \\
& 95.09011808576756 \frac{\mathrm{mi}}{\mathrm{hr}} \\
& \mathbf{9 5 . 1} \frac{\mathrm{mi}}{\mathrm{hr}}
\end{aligned}
$$



# Adding and Subtracting 

 MEASURED numbers.$$
35.2 \mathrm{~cm}+3.6152 \mathrm{~cm}
$$

Degree of Precision
Which number is more precise?
$1 \pm 0.5$


## Adding and Subtracting MEASURED Numbers

A chain is only as strong as its weakest link!


$$
\begin{aligned}
& =5.21-0.083+87.5 \\
& =92.627=92.6
\end{aligned}
$$

## Using Scientific Notation to

 Properlly Show Sig Figs.
##  $=1.00 \times 10^{4}$ <br> 3 sig figs

## Review Graphing

1. Identify the variables Independent variable - X axis

Manipulated variable
Factor adjusted by experimenter
Dependent Variable - Y Axis
Responding variable
Depends on the independent variable
Variable that is expected to change

## 2. Determine the variable range

Subtract the lowest data value (usually zero) from the highest data value for each variable

## Review Graphing

## 3. Determine the scale of the graph

Determine the numerical value for each grid unit that best fits the range of each variable

$$
\text { Range }=\text { round to } 1,2,5,10 \text { etc }
$$ \# of Lines

4. Number \& label each axis and title
5. Determine the data points \& plot on graph

Don't use a small dot, it will get lost

## 5. Draw the graph

Draw a curve or a line that best fits the data points. Do not connect the dots
Use a ruler or straight line

## Scientific Graphs




- Most scientific graphs are made as line graphs. There may be times when other types would be appropriate, but they are rare.
- The lines on scientific graphs are usually drawn either straight or curved. These "smoothed" lines do not have to touch all the data points, but they should at least get close to most of them. They are called best-fit lines.
- In general, scientific graphs are not drawn in connect-the-dot fashion.


## Reviewing Graphing

Independent variable - $\mathbf{X}$ axis

- Manipulated variable
-Factor adjusted by experimenter


## Dependent Variable - Y Axis

-Responding variable
-Depends on the independent variable

- Variable that is expected to change

Title: Dependent vs Independent

$$
\mathrm{y} \text { vs } \mathrm{x}
$$



Independent Variable

## Directly Proportional and Inversely Proportional Graphs

## Directly Proportional



As the independent variable increases (X), the dependent variable ( Y ) increases as well.

## Inversely Proportional



As the independent variable increases ( x ), the dependent variable decreases (Y).

## How to Construct a Line Graph

## 1. Identify the variables

- Independent variable
- Dependent variable

2. Determine the scale of the Graph

- Determine Range - Highest value on data table minus lowest(or Zero)
- Determine Scale (numerical value for each square) that best fits the range of each variable


Independent Variable

- Unless there's a good reason, plot from $(0,0)$
- Choose easy to work with scales (multiples of $2,5,10$ ) and make the graph as large as possible


## How to Construct a Line Graph

3. Number and Label Each Axis

- This tells what the lines on your graph represent. Label each axis with appropriate units


## 4. Plot the Data Points

- Make data points obvious. Small dots get lost.


## 5. Draw the Graph

- draw a curve or line that best fits the data points.


## 6. Title the Graph

- Your title should clearly tell what the graph is about
- If your graph has more than one set of data, provide a key to identify the different lines.


## Predicting Data on a Graph




- Graphs are a useful tool in science. The visual characteristics of a graph make trends in data easy to see.
- One of the most valuable uses for graphs is to "predict" data that is not measured on the graph.
- Extrapolate: extending the graph, along the same slope, above or below measured data.
- Interpolate: predicting data between two measured points on the graph.


## Interpolate vs Extrapolate

- Interpolate
- Predicting an unknown data point within the range of the a known (experimented) data set


## - Extrapolate

- Predicting an unknown data point outside of the range of a known data set
- For Both we use a trend (usually an equation from that trend) established from known data set to predict unknown data points, inside or outside of known range


## Linear Relationship

- $y=m x+b$
- The two variables are directly proportional
- $m$ - Slope-rise/run = change in $y /$ change in x
- For linear relationship the Slope more specifically tells the relationship between $x$ and $y$
- b-y-intercept - Point at which the line goes through the $y$-axis



## Inverse Relationship

- $y=a / x \quad$ hyperbola
- The variables $x$ and $y$ are inversely related to each other
- As one goes up, the other goes down



## Quadratic Relationship

- $y=a x^{2}+b x+c$

Parabola

- This is a square relationship
- $y$ is proportional to $x^{2}$


Accuracy and Precision

-     - Accuracy describes how well the results agreed with the standard or accepted values or outcomes
-     - Precision describes how well the results agreed with each other.



## Extra- Uncertainty in Measurement

There are two kinds of numerical data: exact and inexact. Exact data are numbers that are known exactly. Inexact data are numbers that are not known and have a degree of uncertainty. When experiments are carried out, there will always be a degree of uncertainty. Uncertainty of measurement is the doubt that exists about the result of a measurement. There is always a margin of error for any instrument. Usually, the margin of error is expressed as $+/-$, which provides a range that the actual measurement falls within.

Laboratory glassware usually lists the uncertainty directly on the instrument. But just in case, the uncertainty of analog instruments (such as graduated cylinders \& burets) is $+/$ - half of the smallest division. The uncertainty of digital instruments (electronic balances, timers \& thermometers) is $+/$ - the smallest scale division.

Example: A stick that is 30 centimeters with an uncertainty of $+/-1 \mathrm{~cm}$ means that the stick is actually between 29 and 31 centimeters long. Most electronic balances read to 0.01 g , but others (ones used in precise analytical experimentation) read to 0.0001 or better.

## Extra - Mathematical Relationships

- Certain relationships always exist between certain variables. A large part of physics is understanding and examining these relationships between different physical quantities.
- *** Remember--- If $y$ and $x$ are our two variables then the ' $y$ ' is always the response to whatever ' $x$ ' does
- In other words, ' $y$ ' is a function of ' $x$ '.
- However, in real physics problems these will not always be $x$ 's and $y$ 's, you will need to determine what is your ' $x$ ' and what is your ' $y$ '

